



To recuperate or not to recuperate - ORC cycles compared to ideal cycles

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UNIVERSITY OF ICELAND
FACULTY OF INDUSTRIAL ENGINEERING,
MECHANICAL ENGINEERING AND COMPUTER SCIENCE



- The performance of real power cycles for heat source in the temperature range from 100°C to 300°C is studied in this paper.
- A reference is made to three ideal power production cycles:
 - Carnot
 - Triangular
 - Lorenz.





- The source fluid flow is 1 kg/s for all calculations
- The real cycles are assumed to have infinite heat exchanger area
- Pumps and turbines are isentropic
- The cooling fluid is produced by external means, without being a parasite of the power plant.





- The binary cycles studied are:
 - ORC with a single high pressure level
 - ORC with two pressure high levels
 - Saturated Kalina cycle
 - Transcritical cycle
 - Single ~~and double flash~~ geothermal power cycles are included as well
 - A few different working fluids are considered for the ORC cycles
 - ~~A few different ammonia concentrations for the Kalina cycle.~~





- The results are presented as:
- $$z = f(T_{source}, T_{return})$$
- The produced power for these cycles from the same source is then compared and a range of superiority for each cycle presented.
- The effect of recuperation on the produced power as well as on the calculated efficiency is shown.





- ~~• Finally the influence of finite heat exchanger area is analyzed and an estimate of the cooling fluid generation parasitic power made both for air and wet cooling tower system.~~





Source properties – the denominator

Geothermal

$$\dot{m}_{source} = constant$$

$$\dot{Q}_{in} \neq constant$$

$$\dot{Q}_{available} > \dot{Q}_{in}$$

$$\eta_{max} = \lim_{\dot{Q}_{in} \rightarrow 0} \left(\frac{\dot{W}}{\dot{Q}_{in}} \right); \Rightarrow \dot{W}(\eta_{max}) = 0$$

$$PC = \frac{\dot{W}}{\dot{m}_{source}} = \frac{E}{M_{source}} \left[\frac{kWh}{ton} \right]$$

WHR

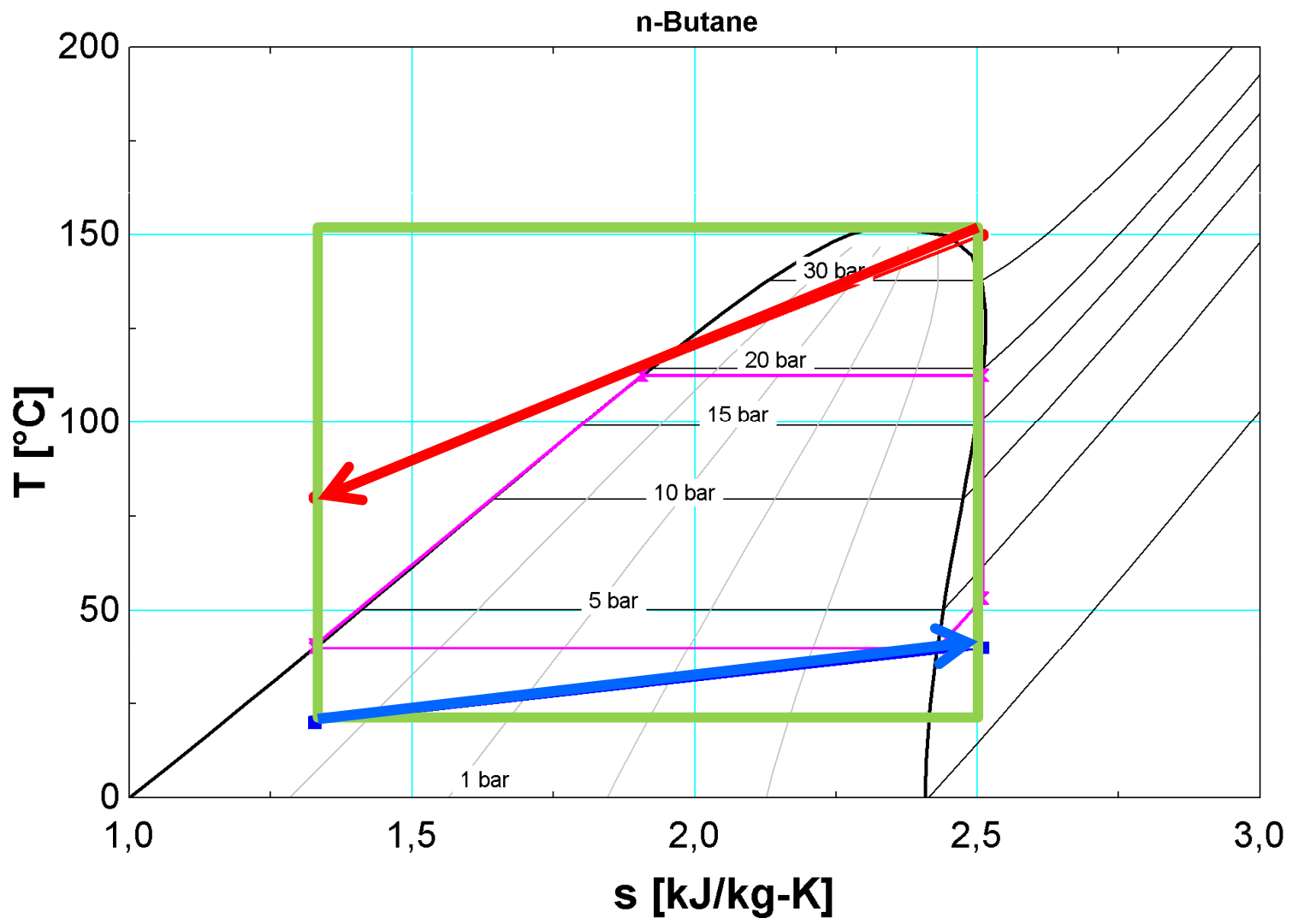
$$\dot{m}_{source} \neq constant$$

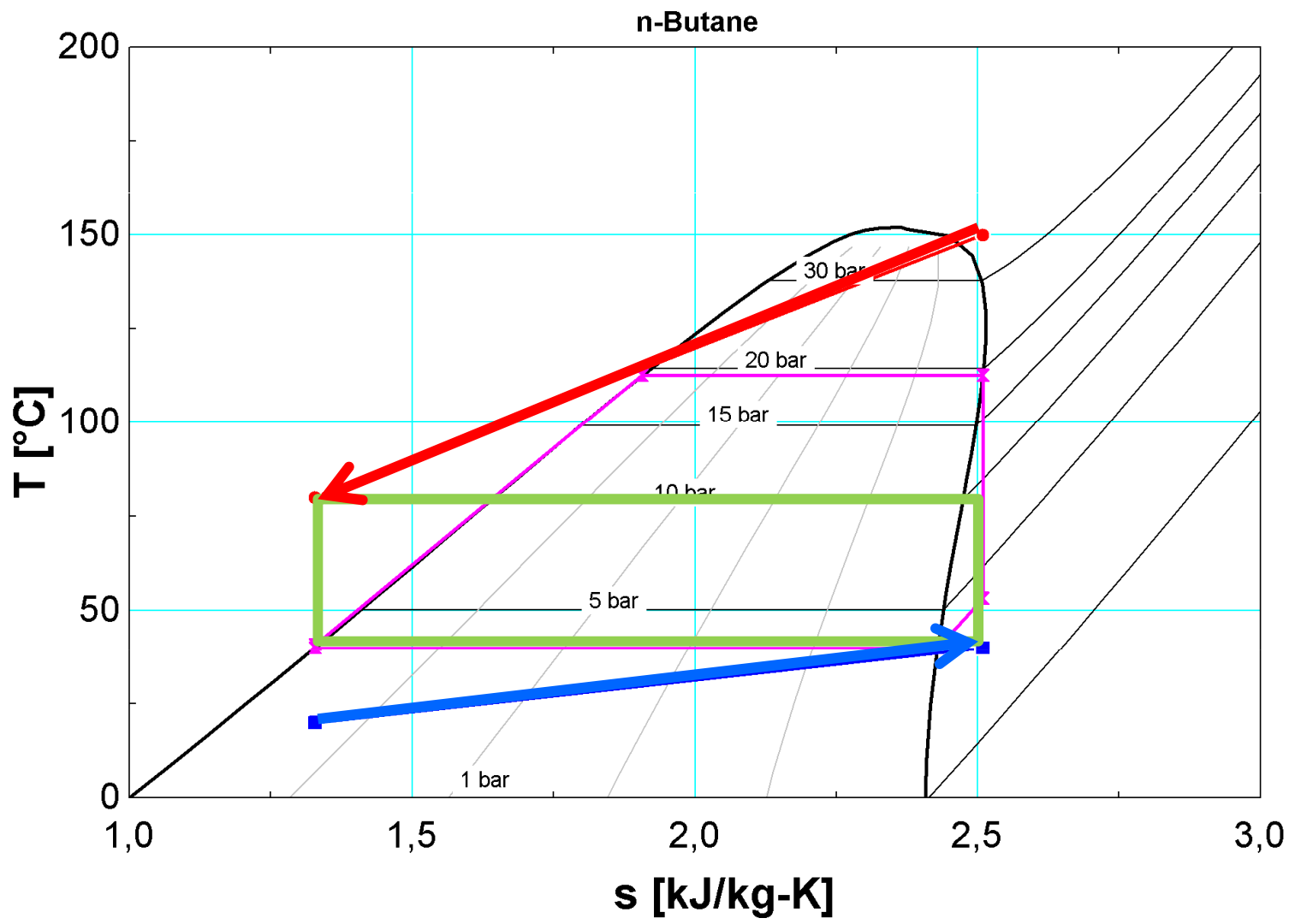
$$\dot{Q}_{in} = constant$$

$$\dot{Q}_{in} = \dot{Q}_{available}$$

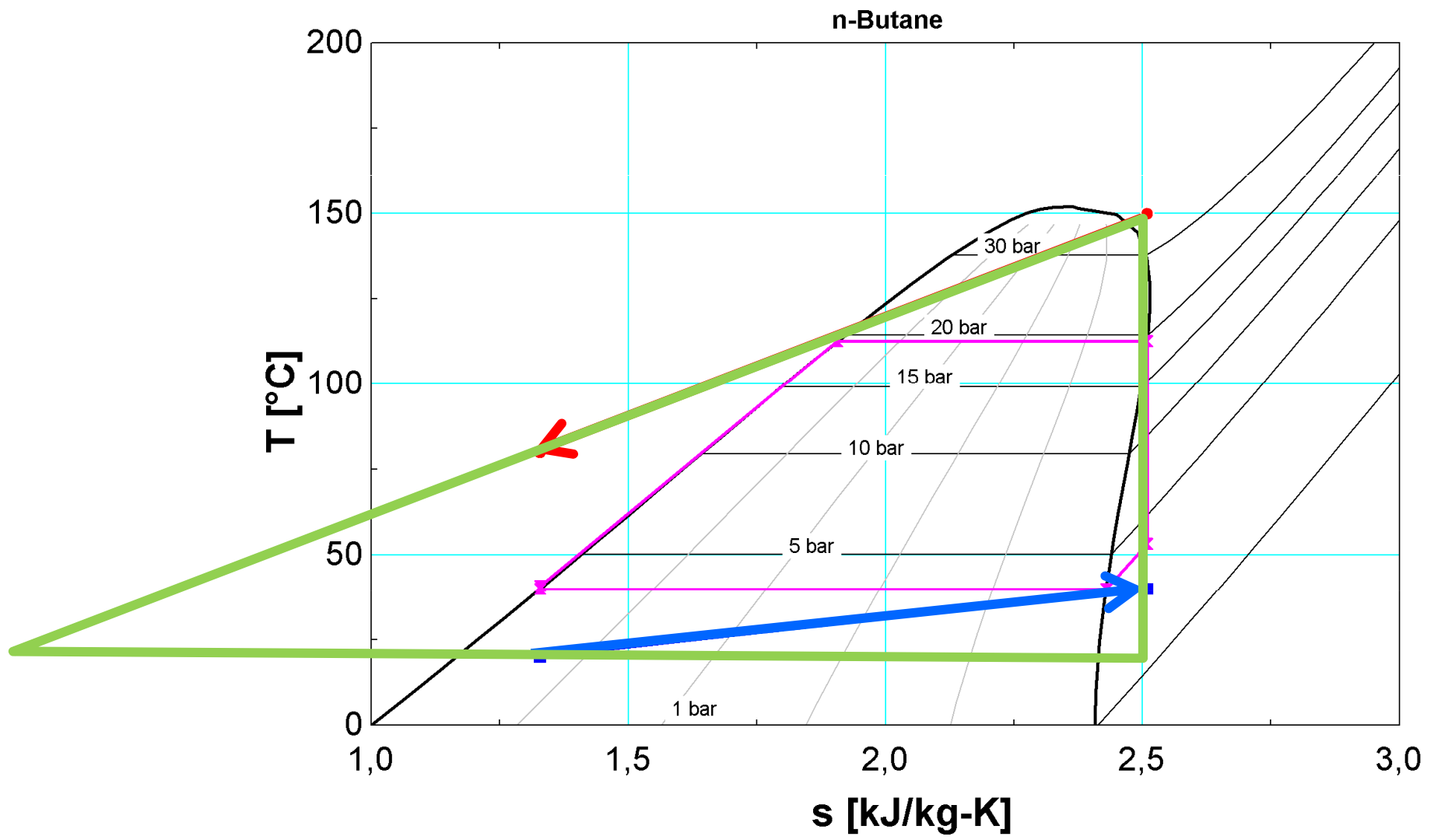
$$\eta = \eta_{max} = \frac{\dot{W}}{\dot{Q}_{in}} = \frac{\dot{W}}{\dot{Q}_{available}}$$

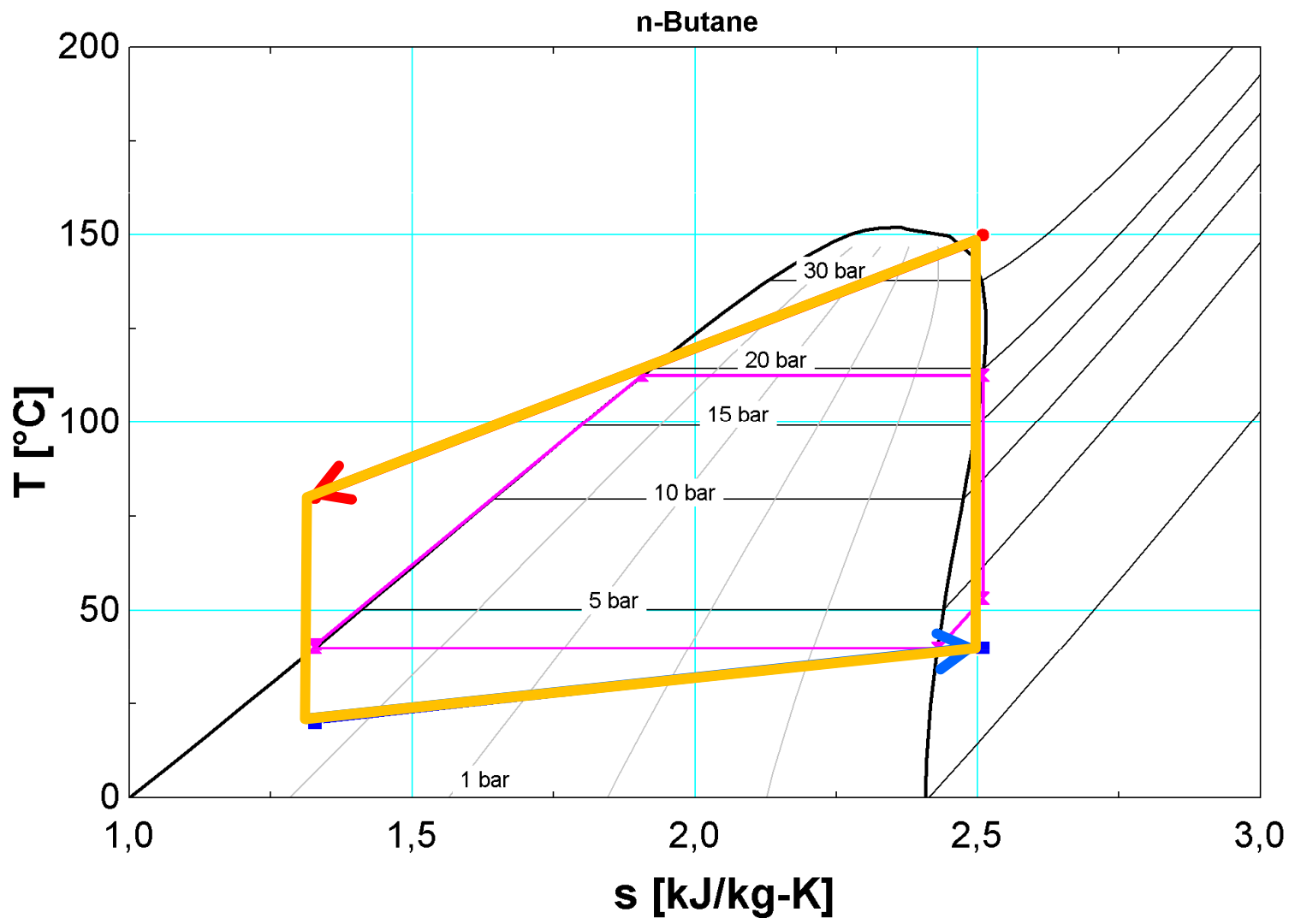


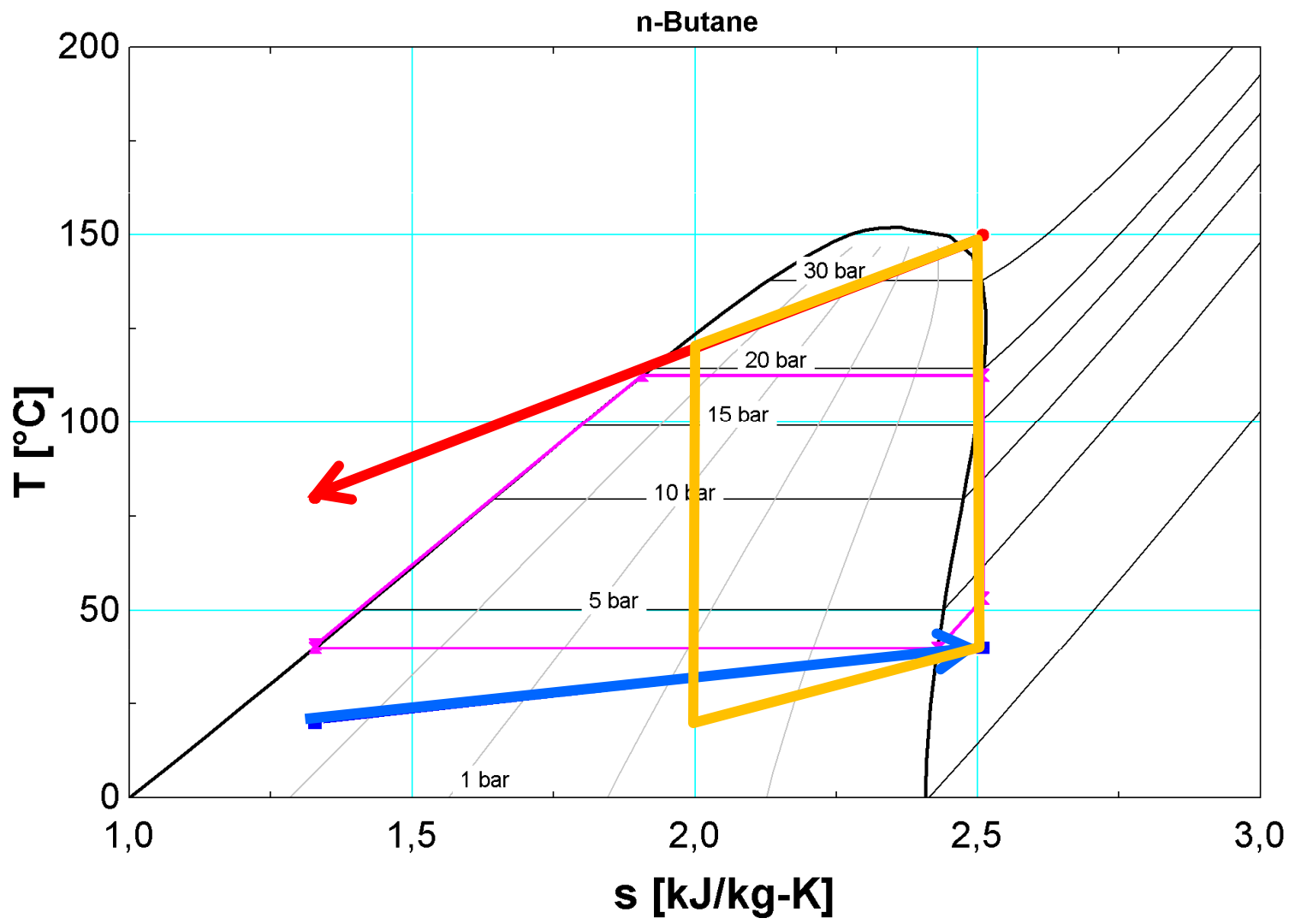


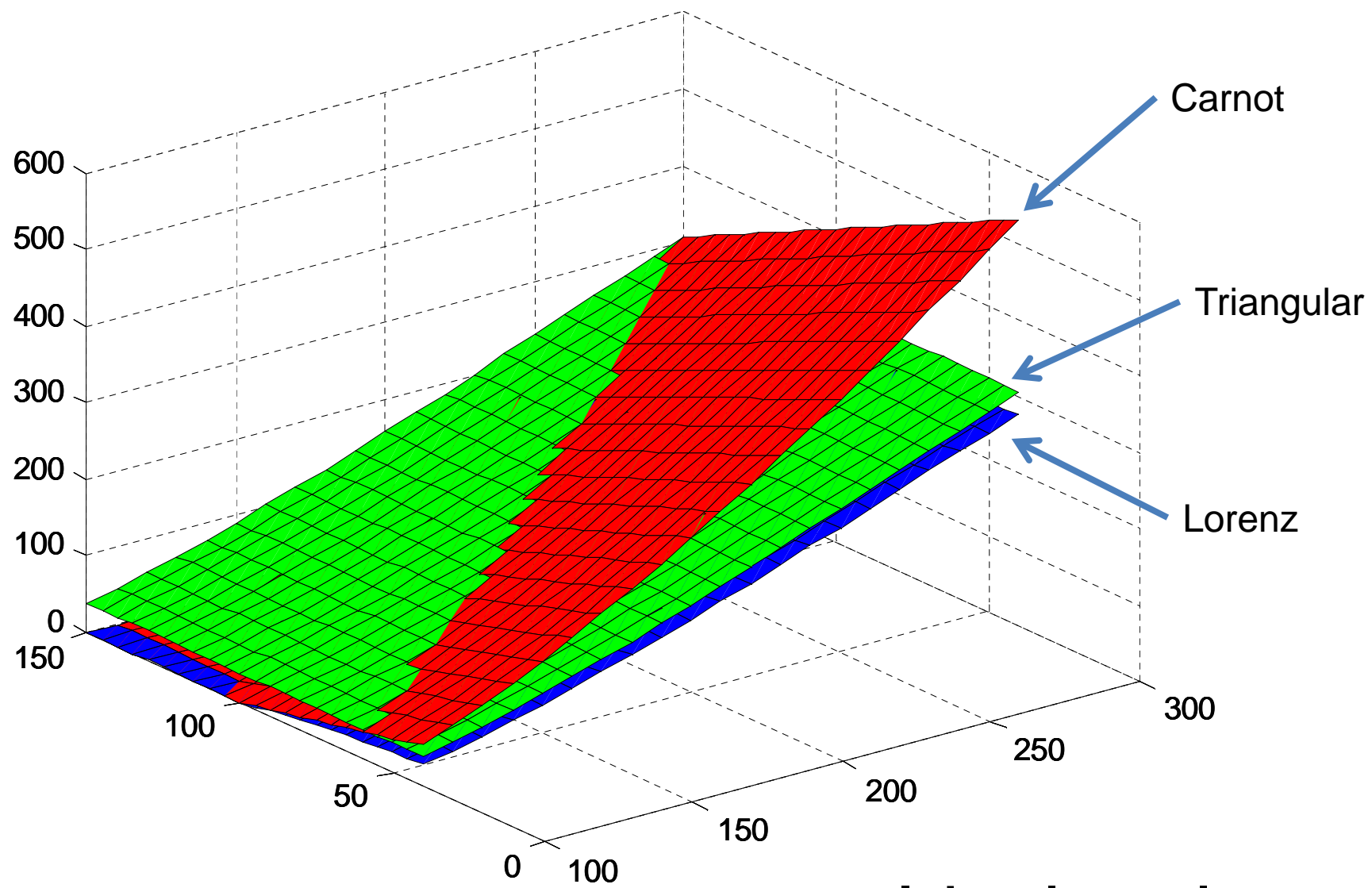


Pessimistic Carnot





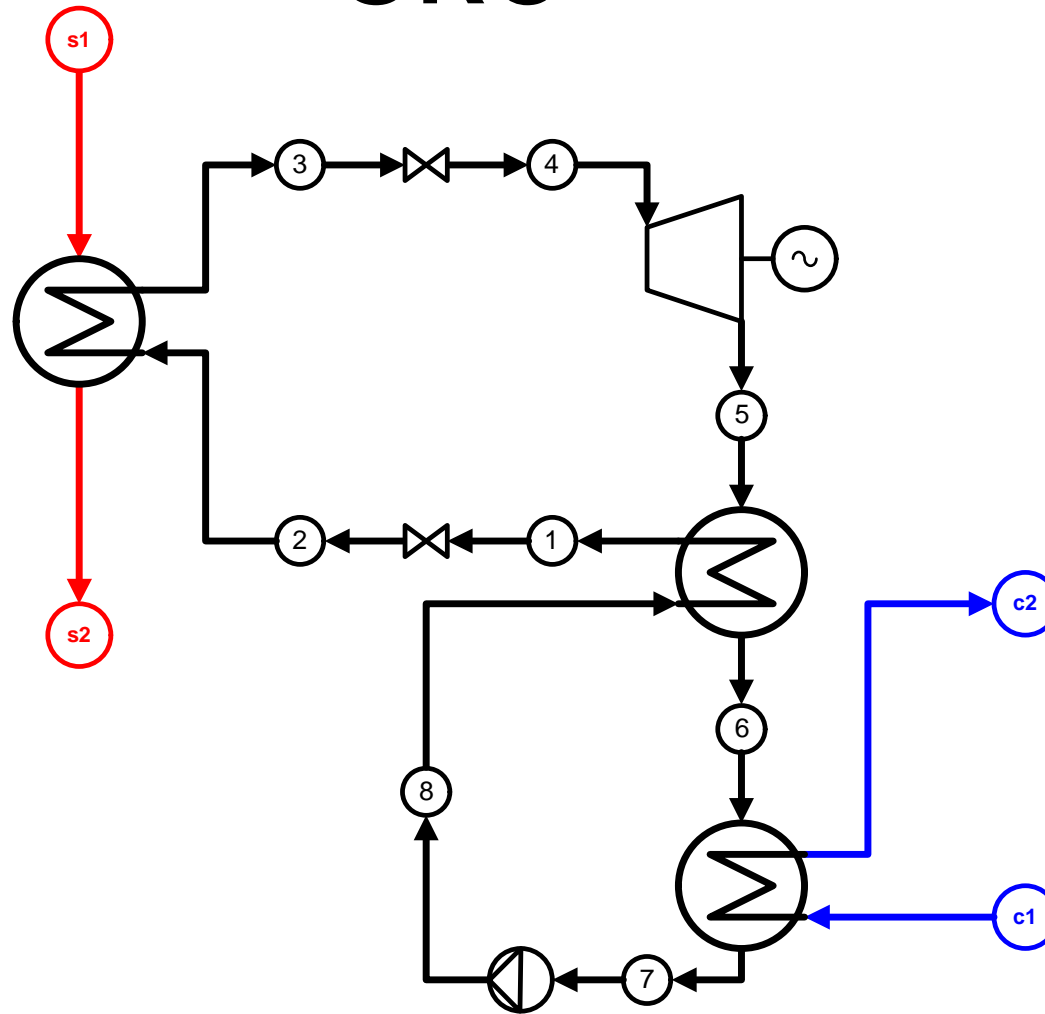


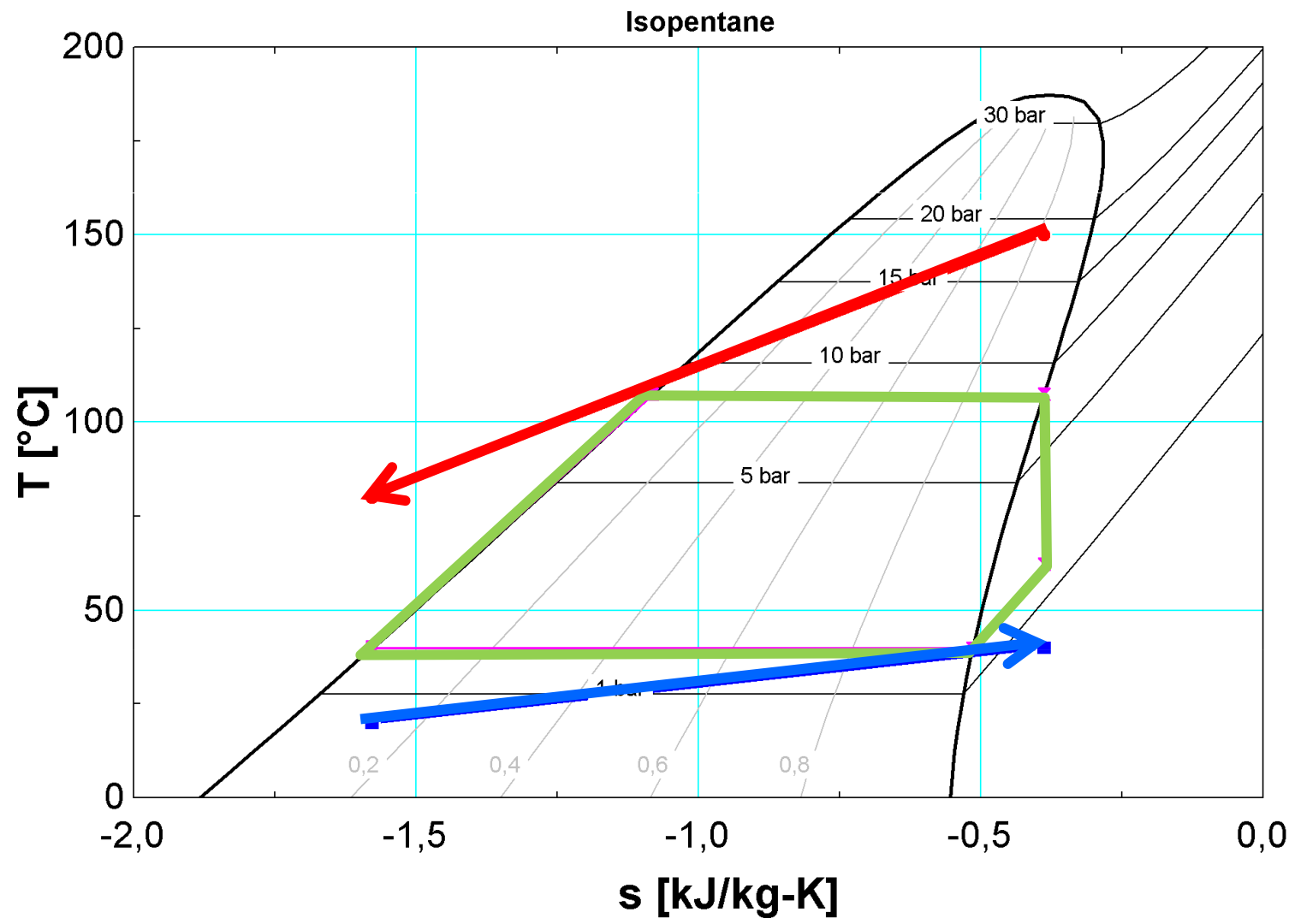


Ideal cycles

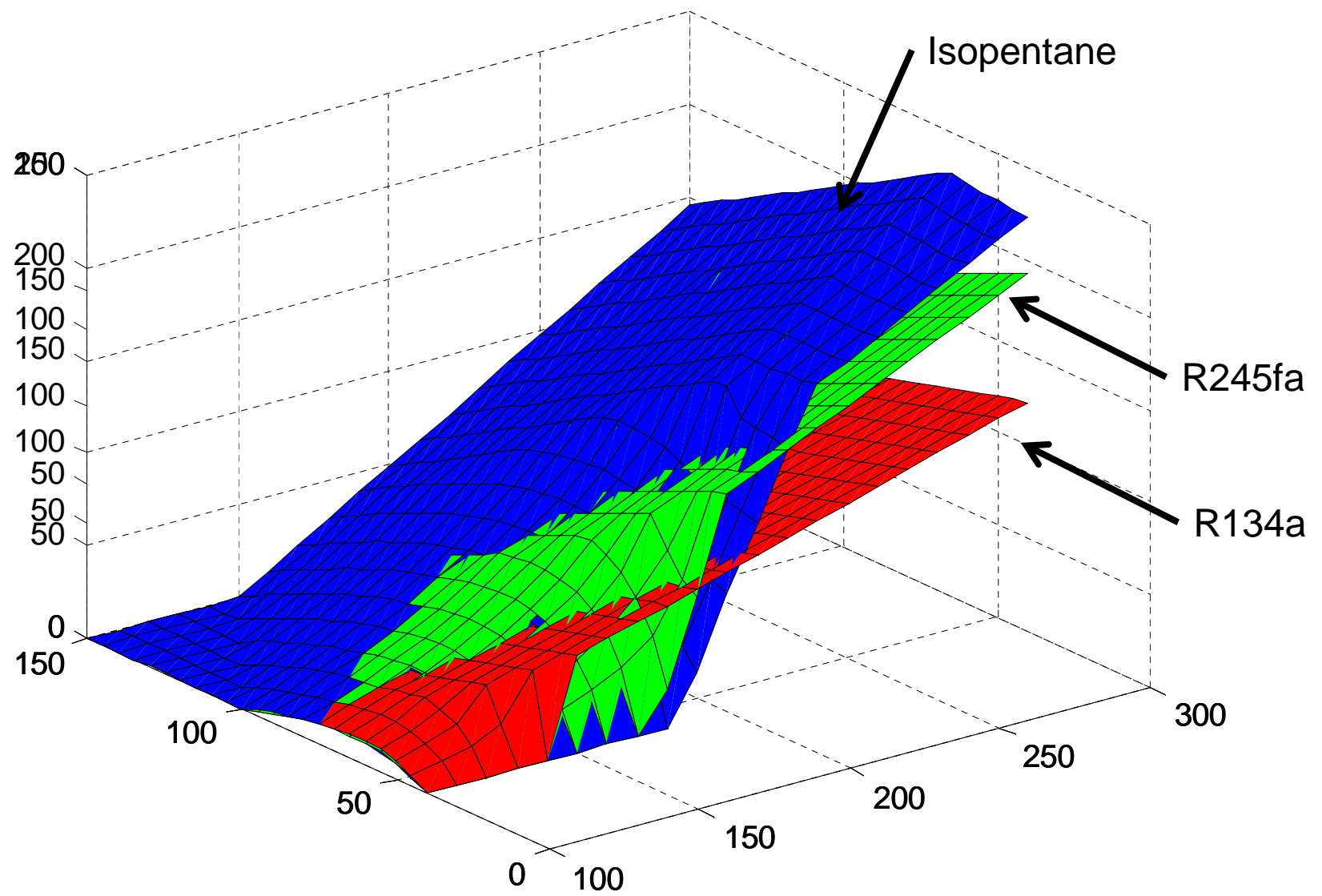


ORC

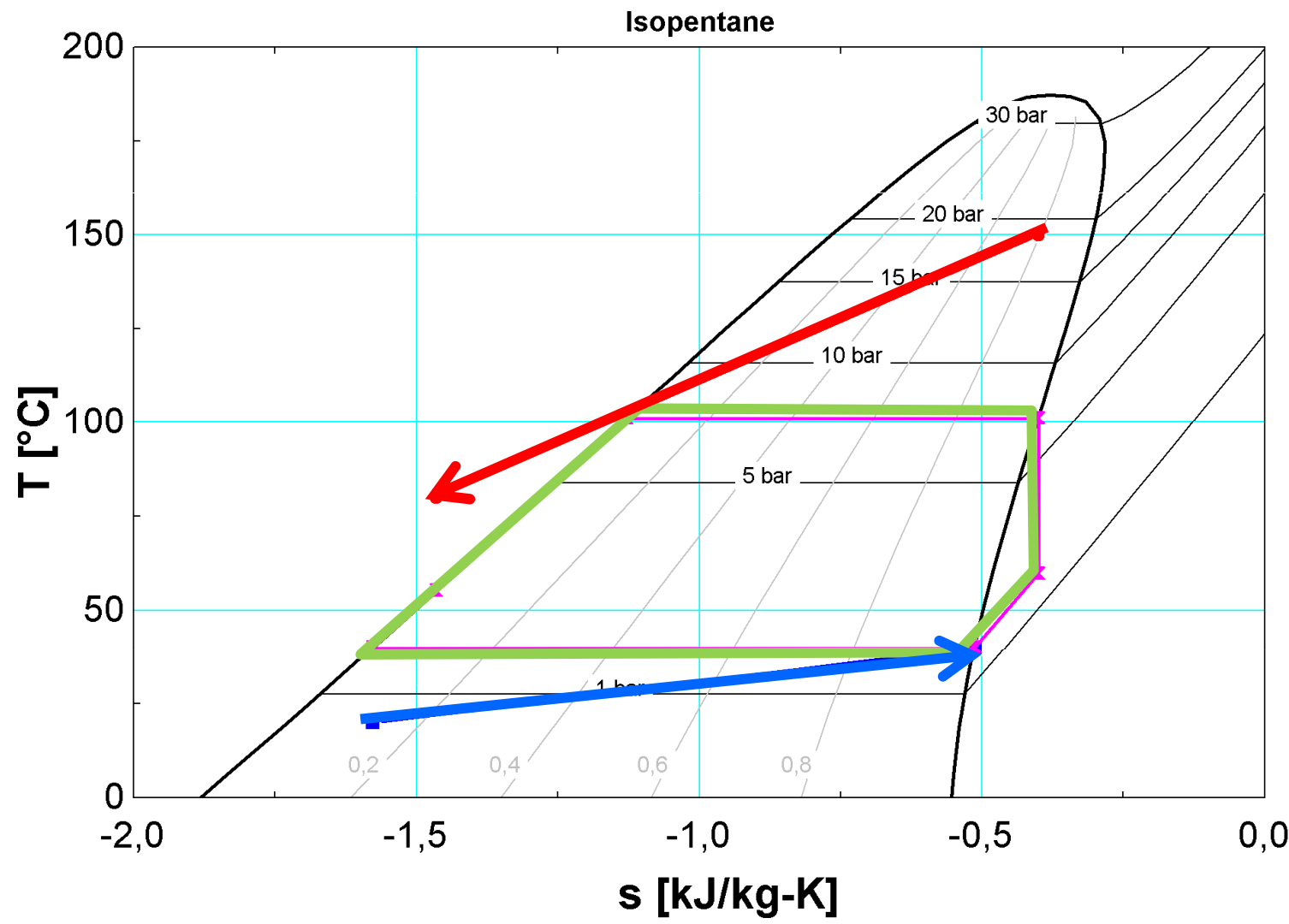


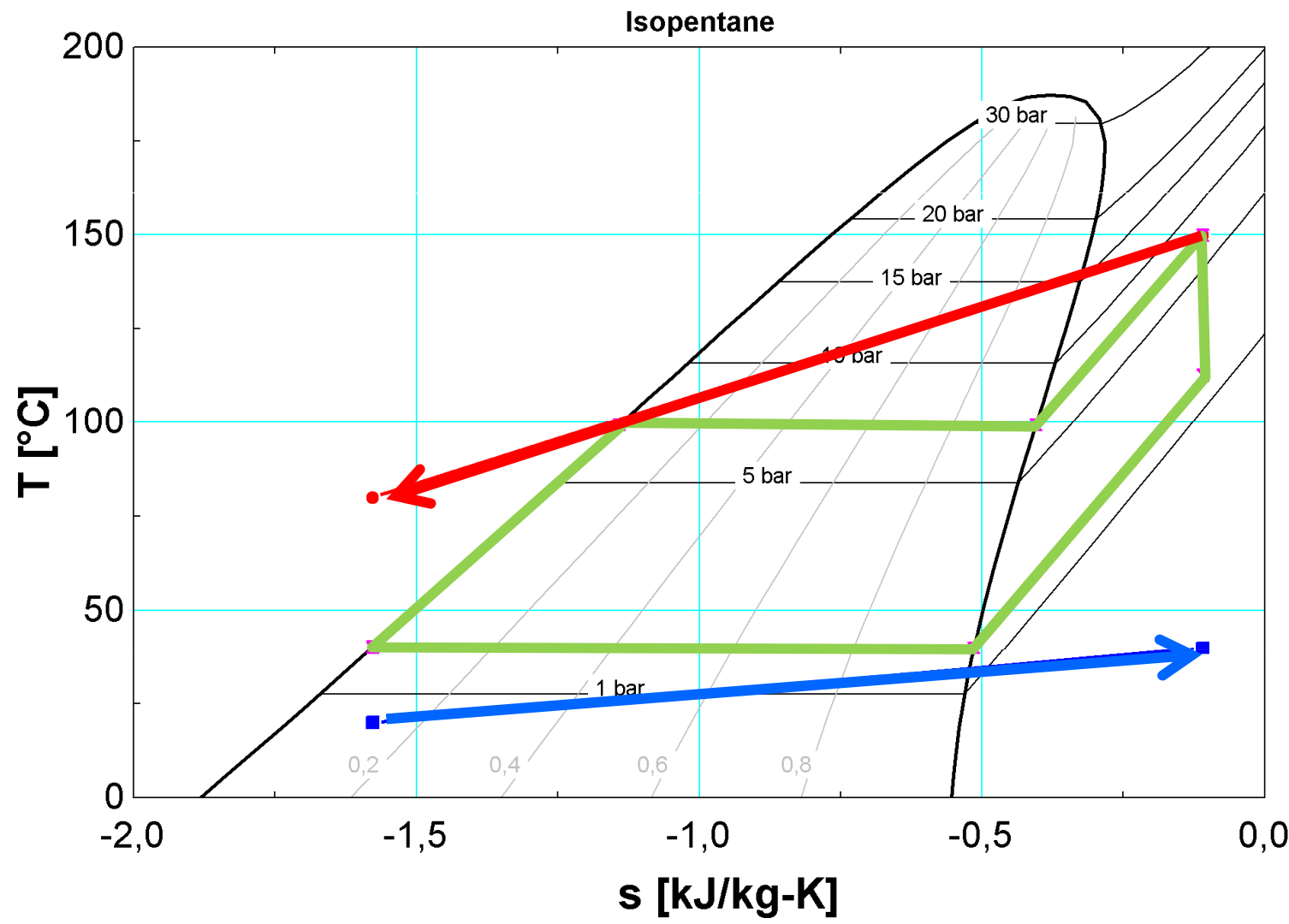


Saturated Isopentane

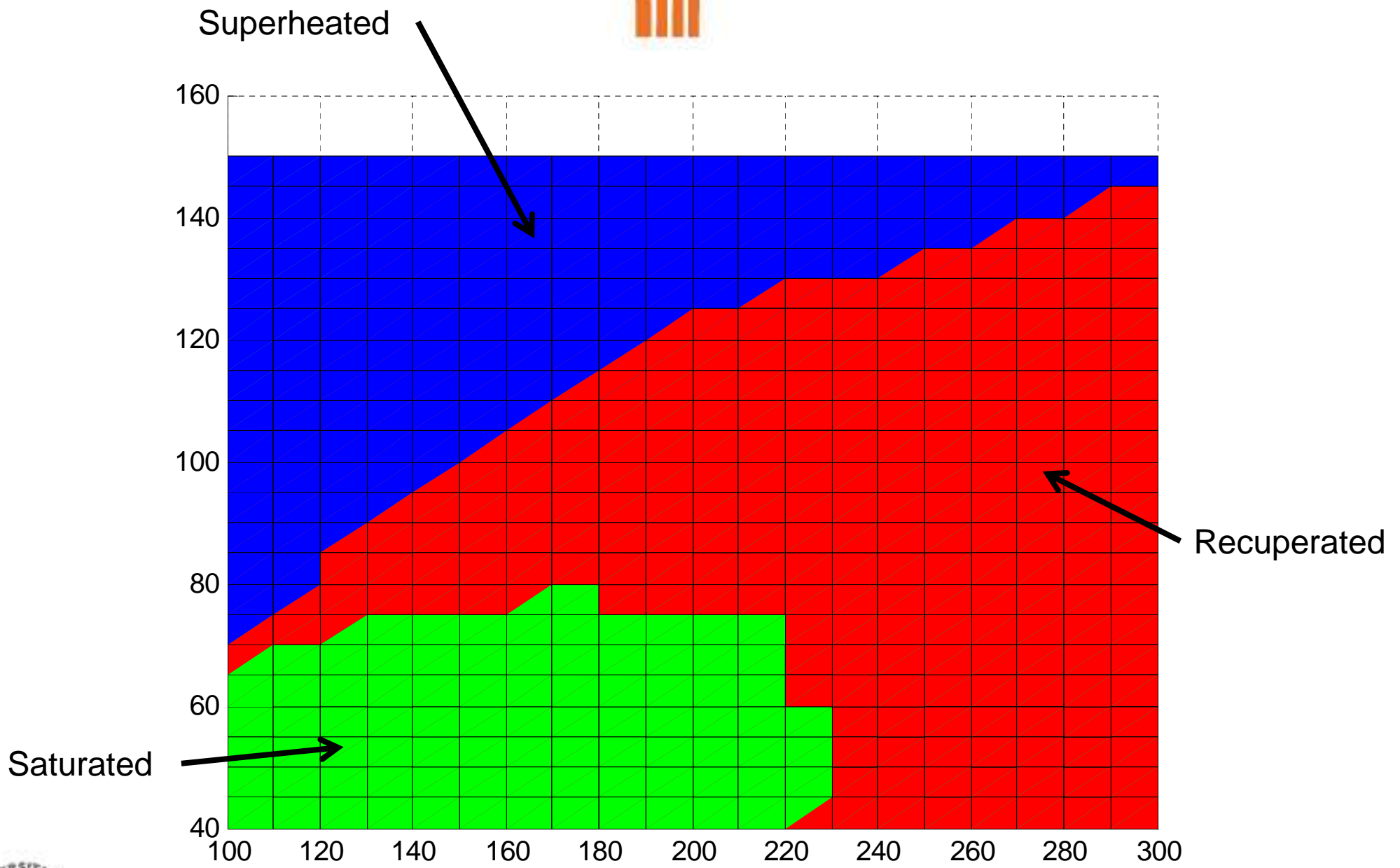


Saturated comparison

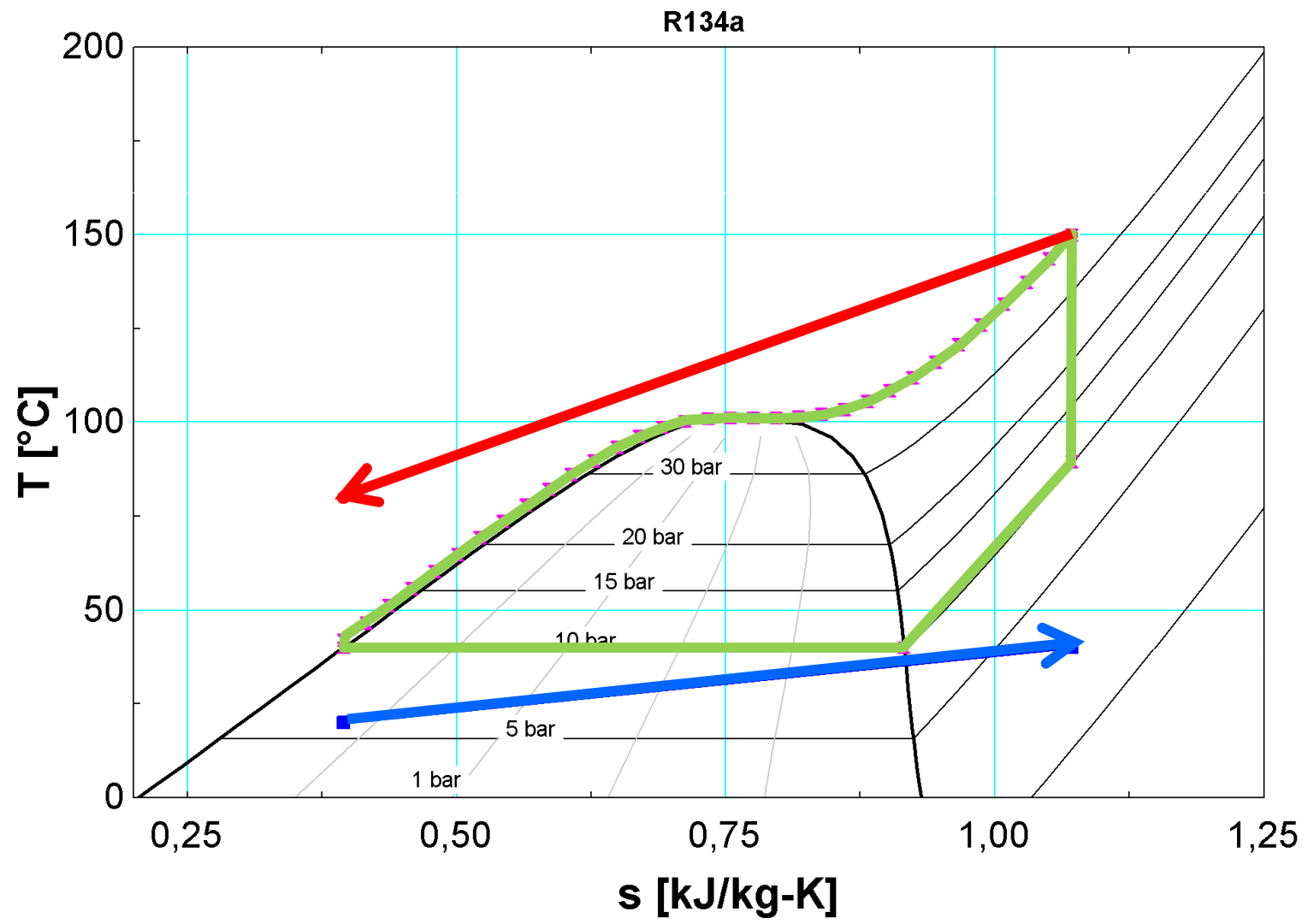




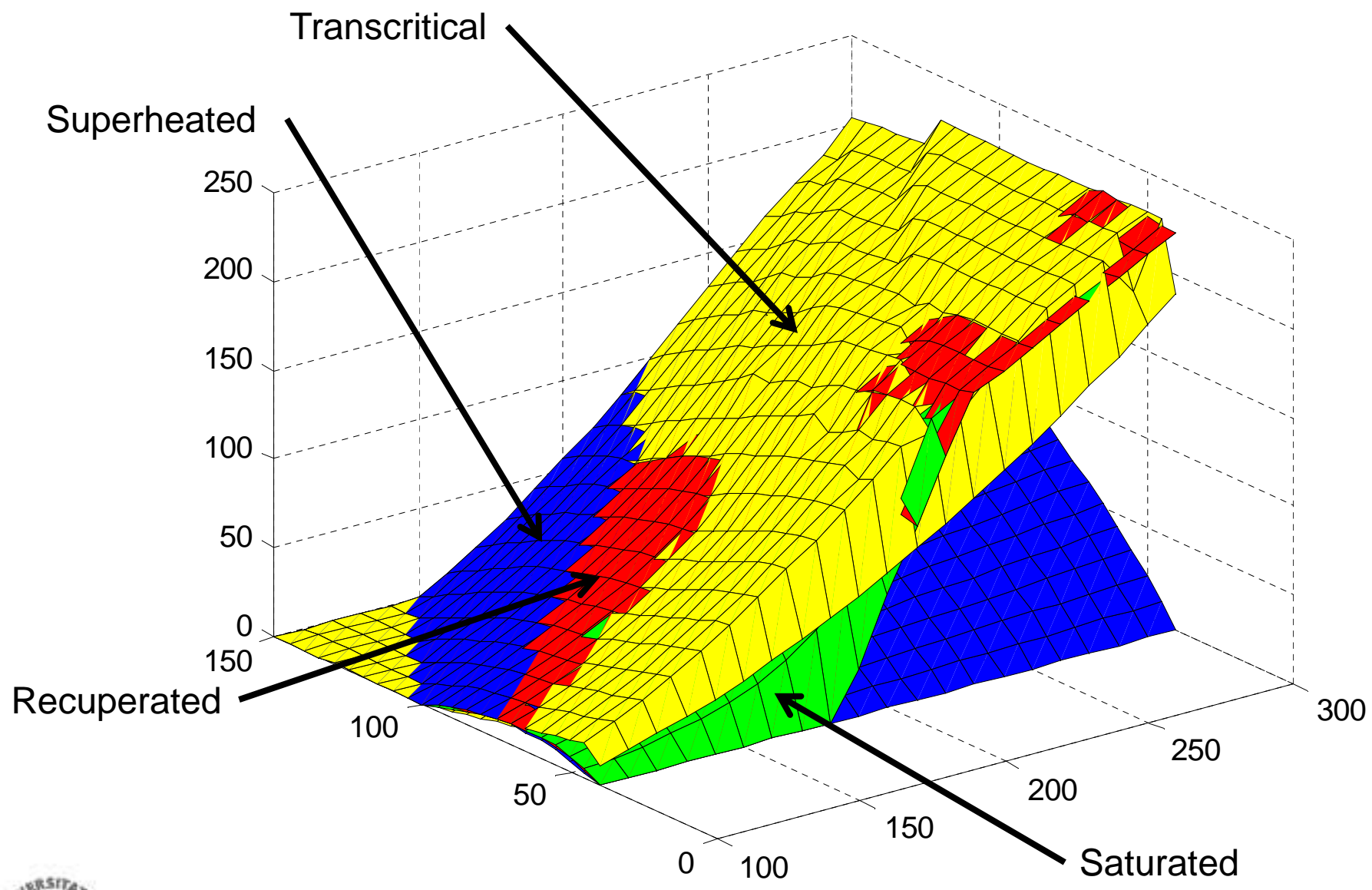
Superheated Isopentane



Superiority ranges

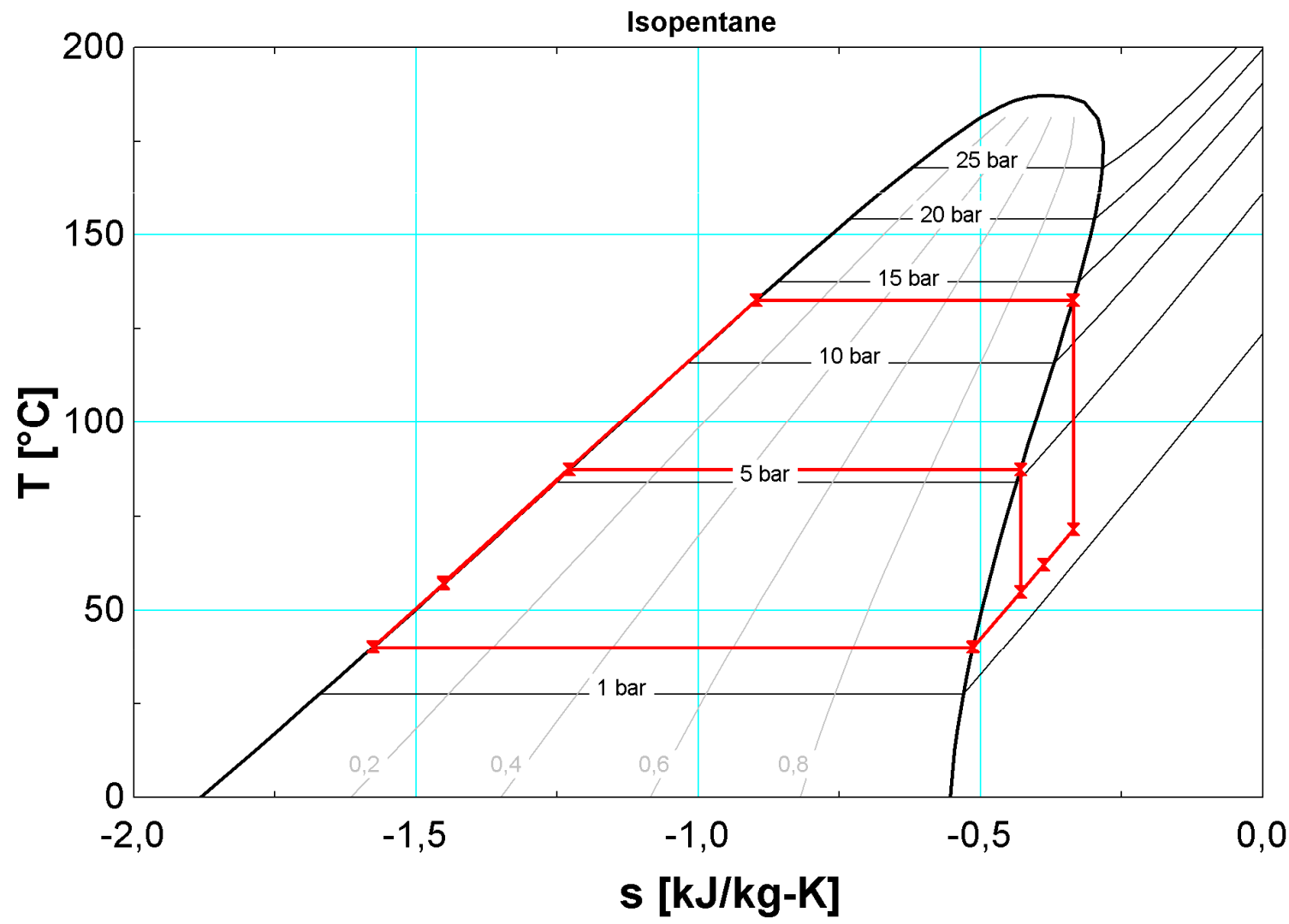


Transcritical Isopentane



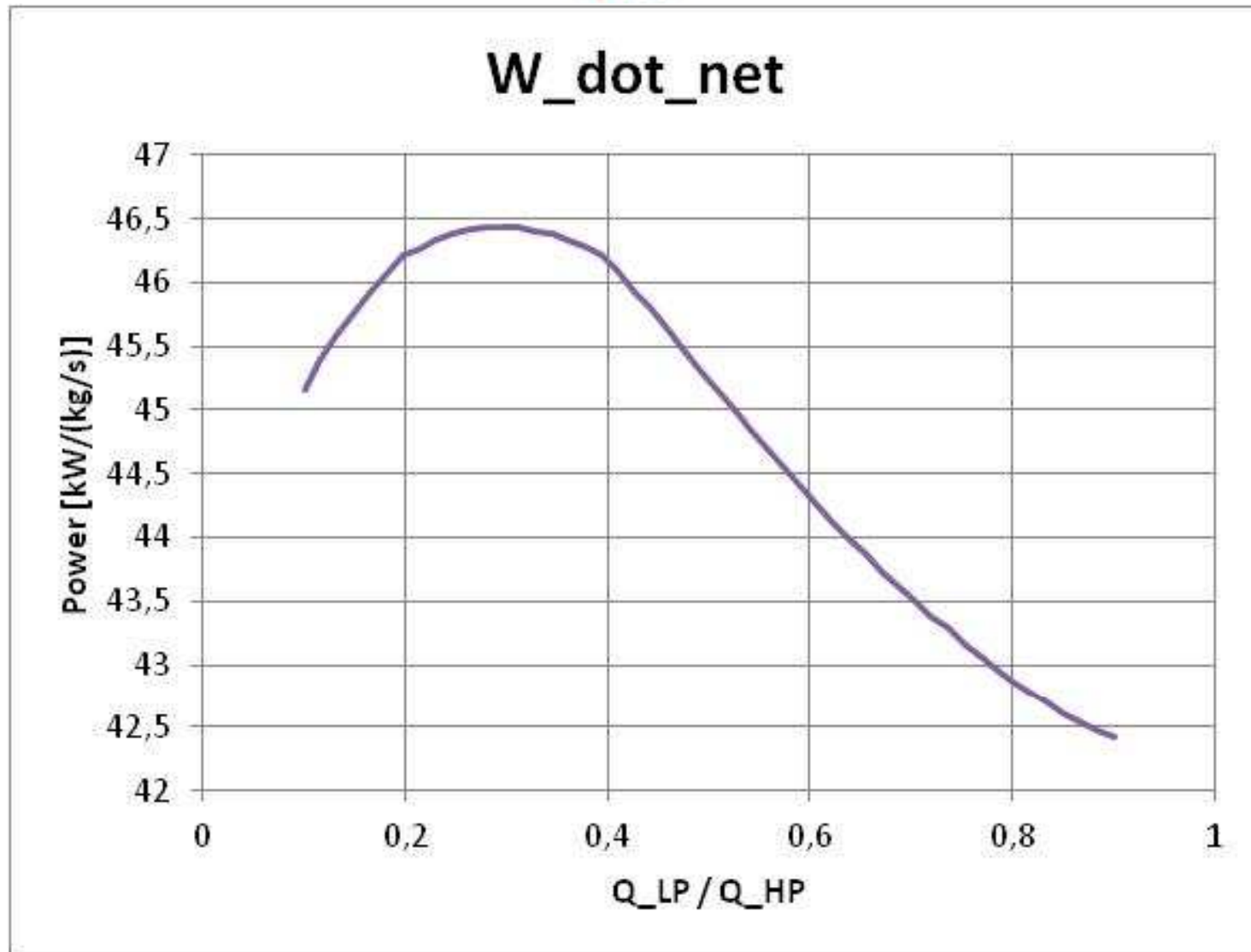
Superiority





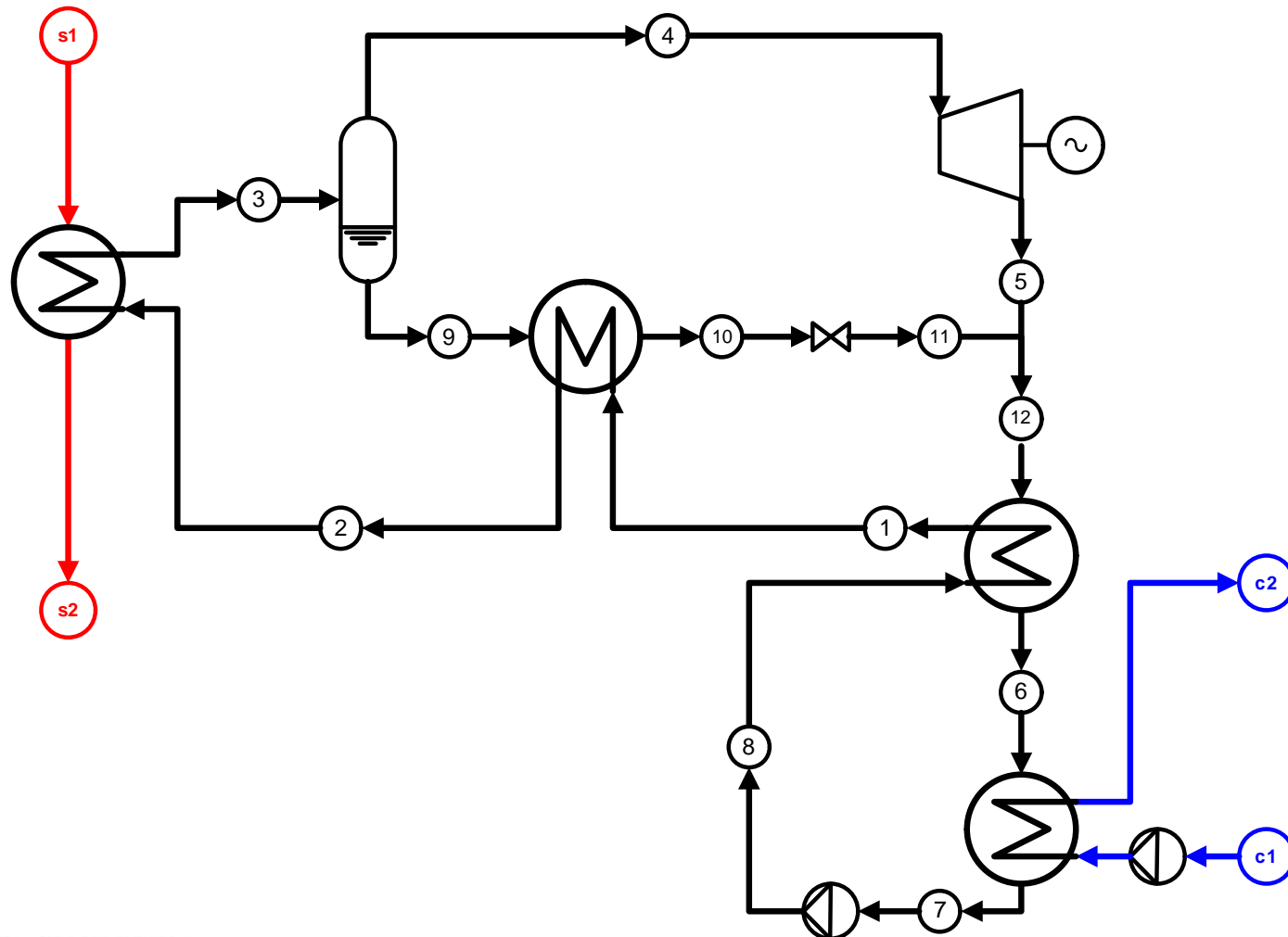


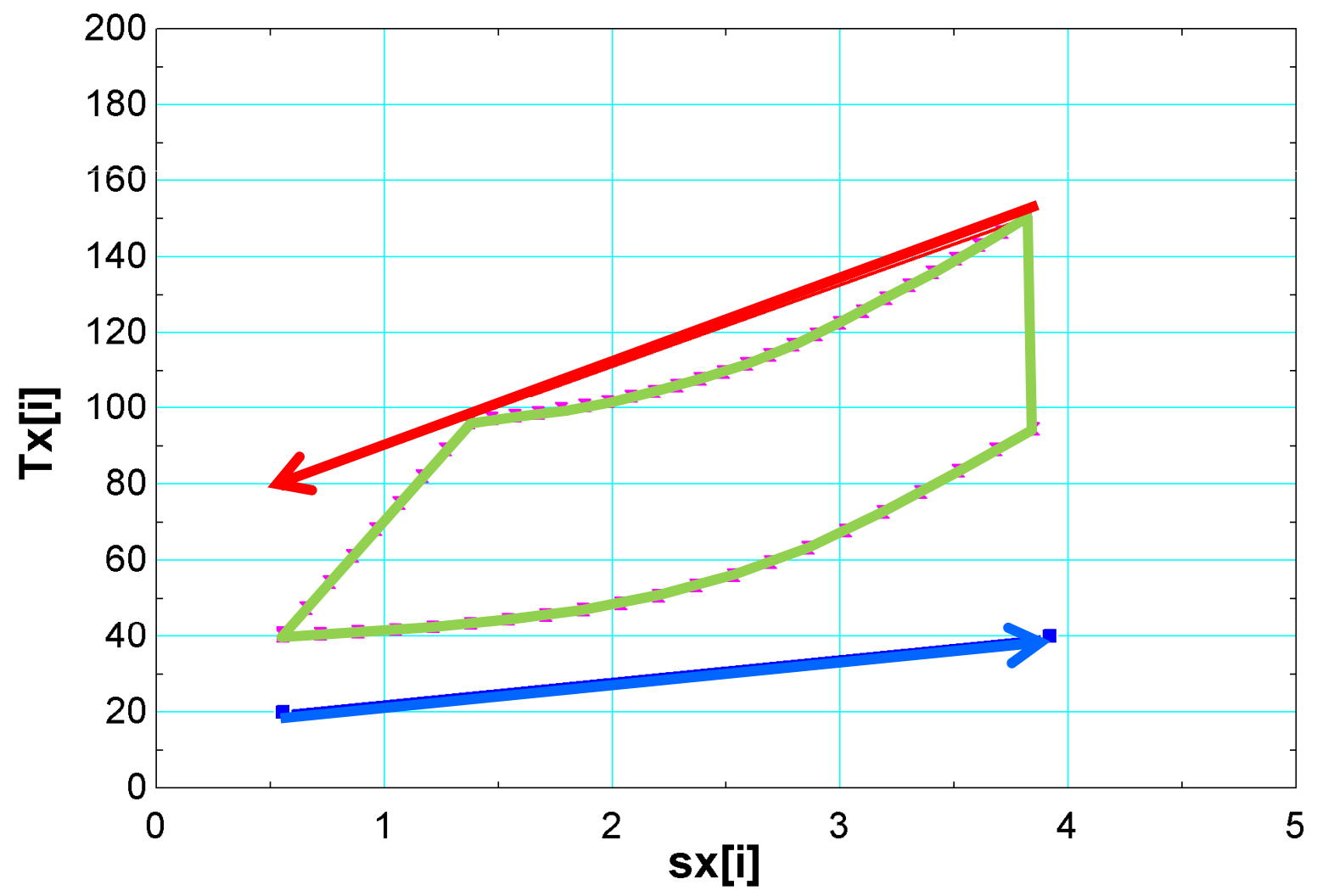
W_dot_net

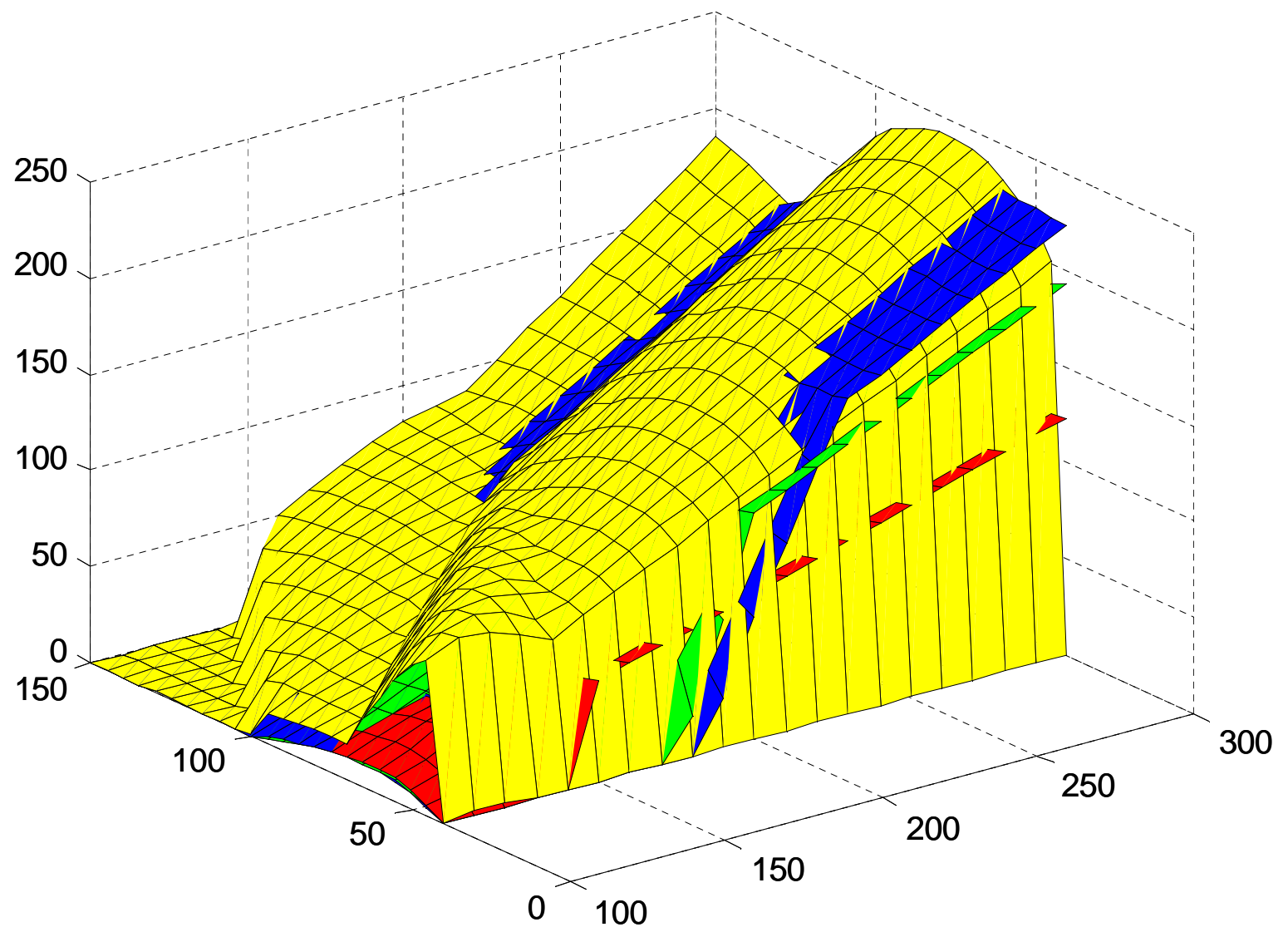




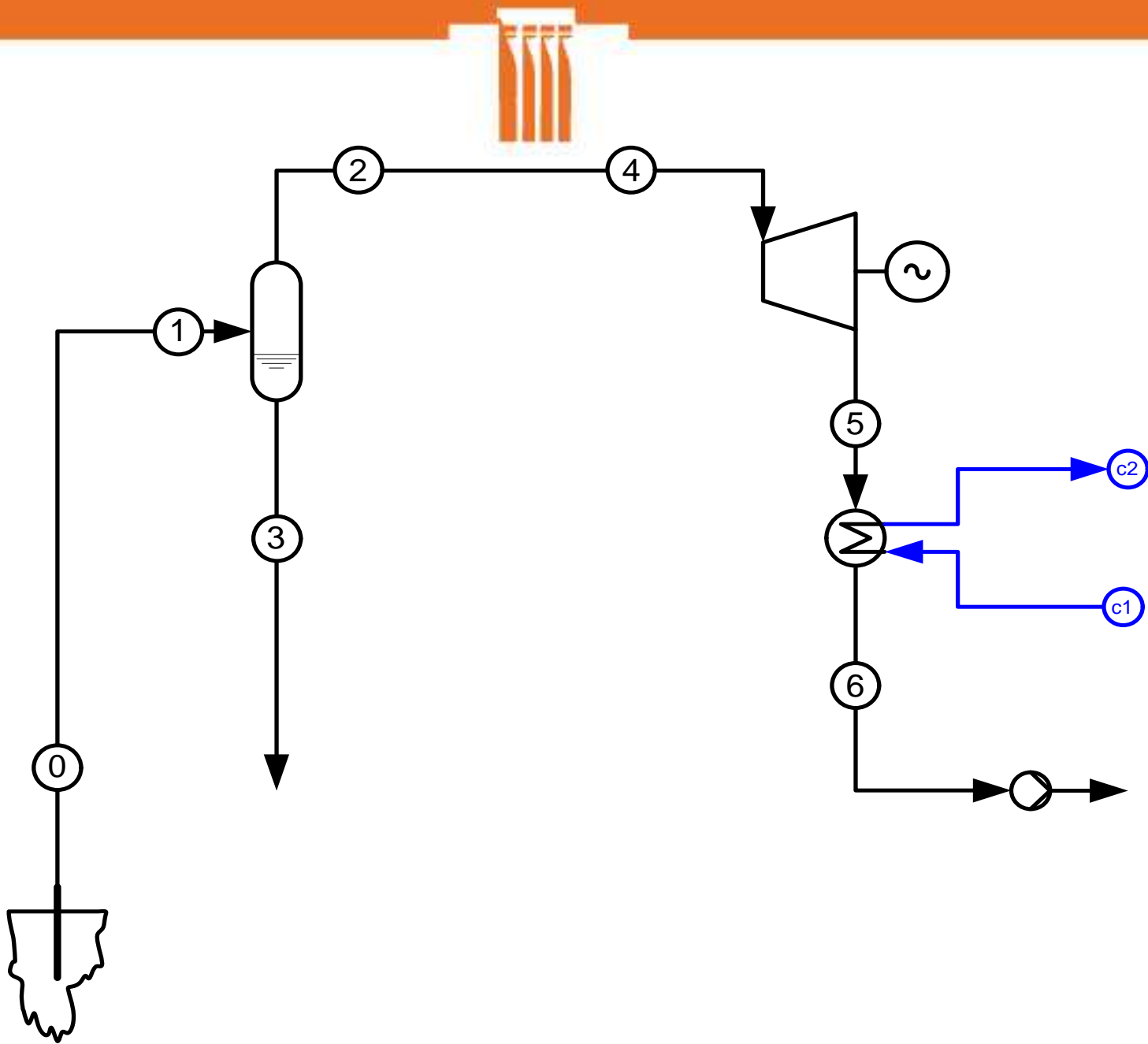
Kalina





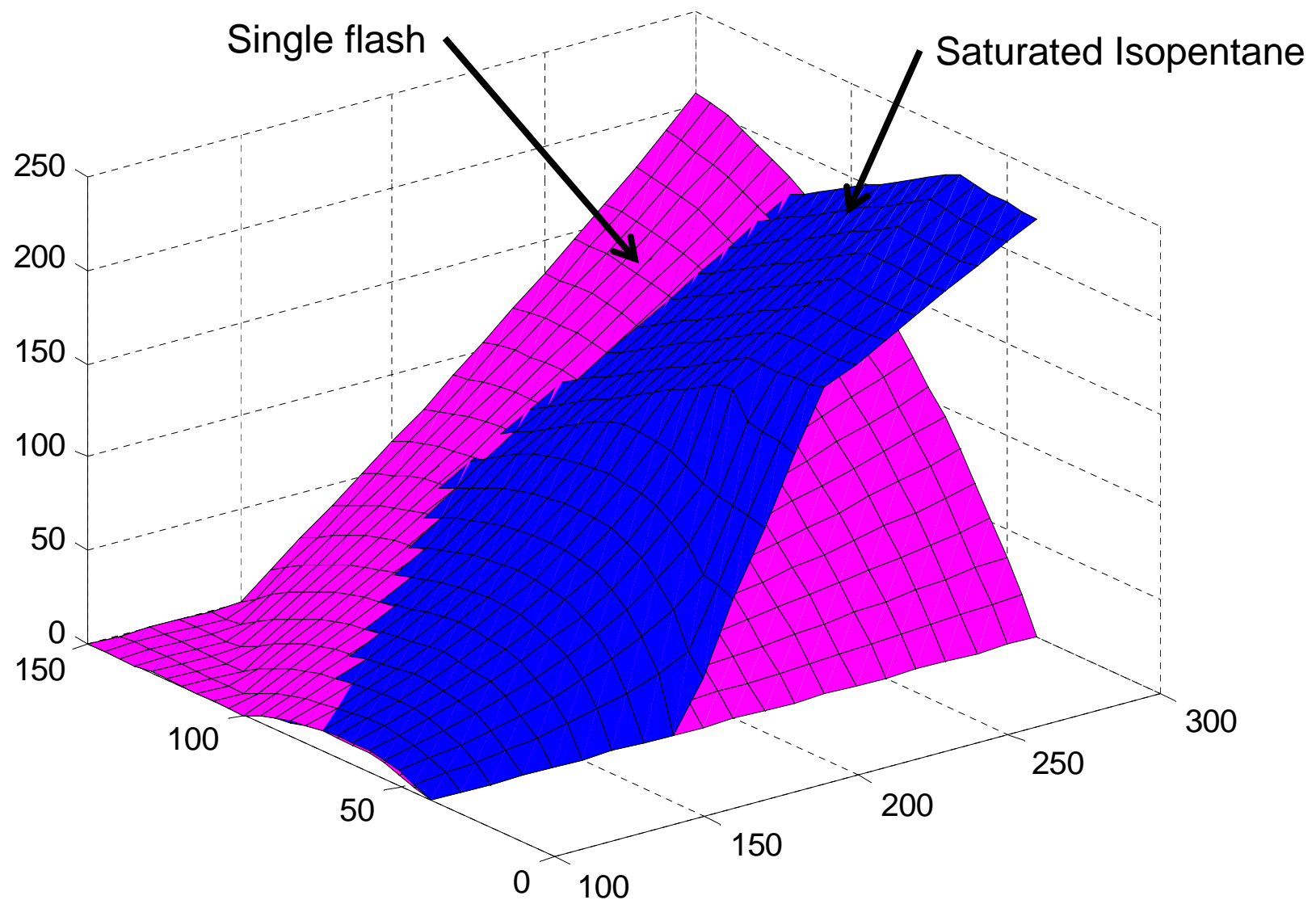


Kalina compared



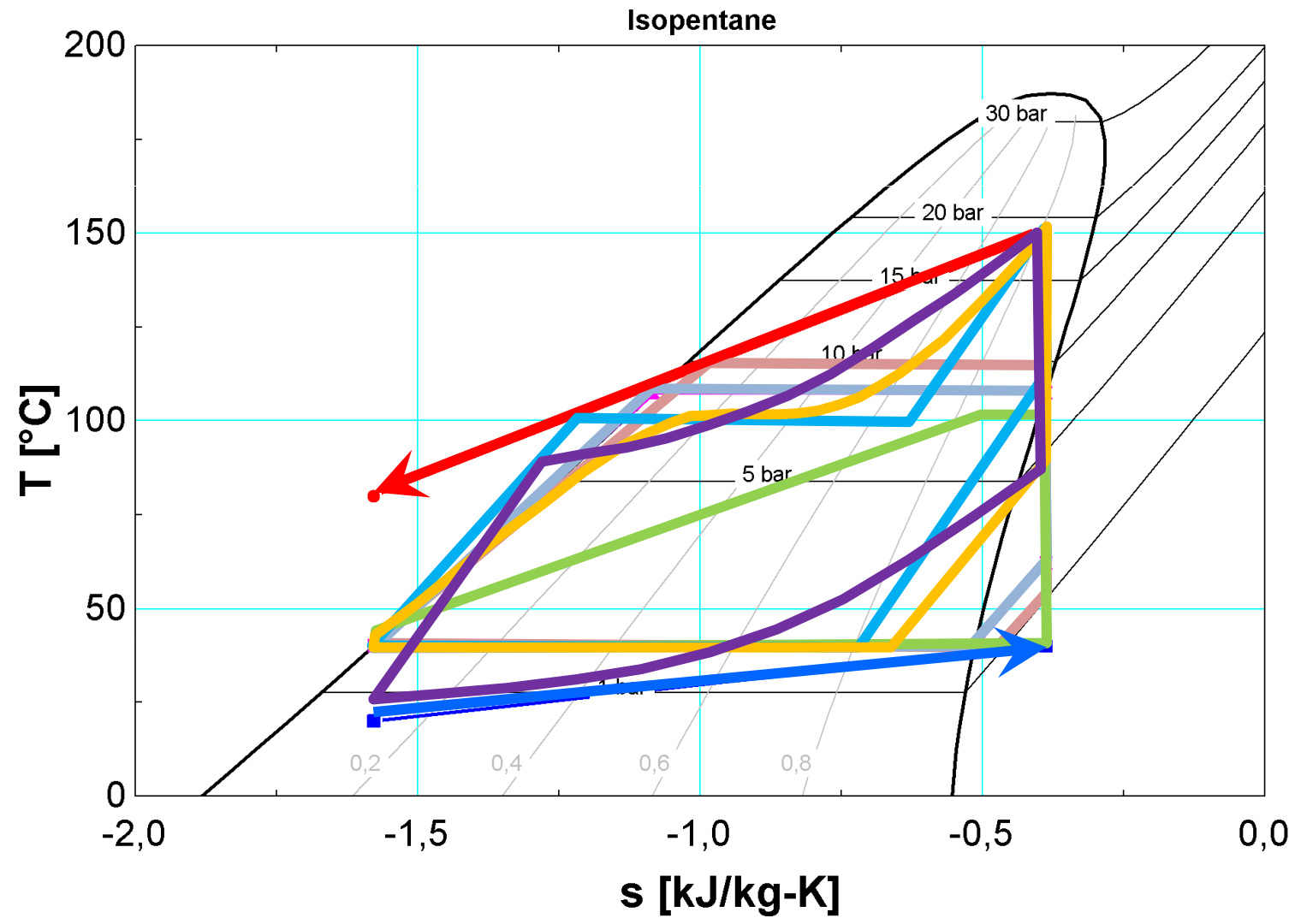
Single flash





Sf compared





- Source
- Sink
- Isopentane
- n-butane
- R134a
- IP superheat
- IP transcritical
- Kalina





Source

Sink

Isopentane

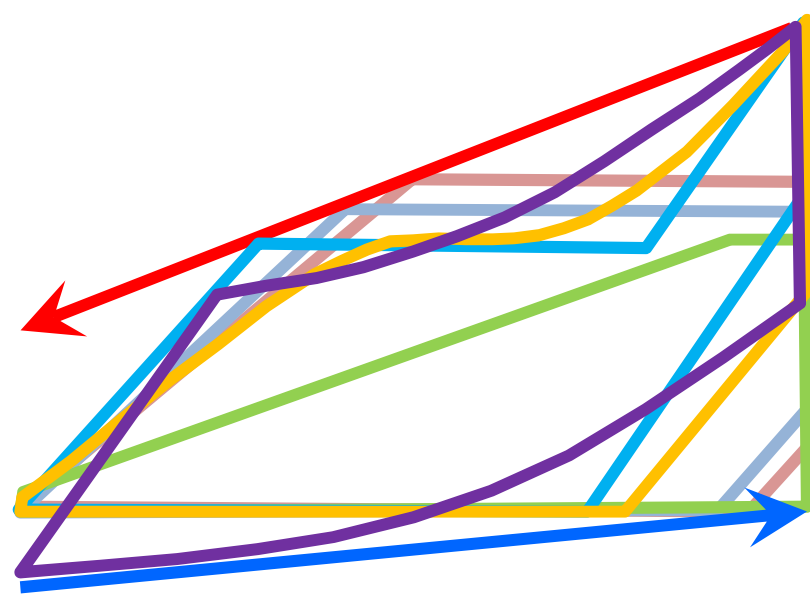
n-butane

R134a

IP superheat

IP transcritical

Kalina





Dank U wel !!

